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- (4) Electrostatic image developing toner.
- An electrophotographic toner containing a compound of the following formula (1):

wherein X is a

(wherein A is an electron attractive group, and each of R1 and R2 which are independent of each other, is a hydrogen atom, a halogen atom, an alkyl group, a cycloalkyl group, an alkoxy group, an aryl group, an aralkyl group or a hydroxyl group, or R1 and R2 together form a ring),



(wherein A, R₁ and R₂ are as defined above), or

$$\mathbf{A} - \overleftarrow{\bigcirc} \mathbf{R_1} \quad \mathbf{R_2}$$

(wherein A, R₁ and R₂ are as defined above), and Y is a hydrogen atom, an alkyl group or an aryl group.

The present invention relates to an electrophotographic toner containing a certain specific compound.

In an image-forming process by means of an electropholographic system, an electrostatic latent image is formed on an inorganic photoconductive material such as selenium, a setenium alloy, cadomiun sulfide or amorphous silicon, or on an organic photoconductive material employing a charge-generating material and a charge-transporting material, and the latent image is developed by a toner, then transferred and fixed on a paper sheet or plastic film to obtain a visible image.

The photoconductive material may be positively electrifiable or negatively electrifiable depending upon its construction. When a printed portion is remained as an electrostatic latent image by exposure, development is conducted by means of an oppositely electrifiable toner. On the other hand, when a printed profiton is destatisized for reversal development, development is conducted by means of an equally electrifiable toner. A toner is composed of a binder resin, a coloring agent and other additives. However, in order to Impart desired tribocharge properties (such as desired charge up speed, tribocharge level and tribocharge level stability), stability with time and environmental stability, it is common to use a charge-control agent. The properties of the toner will be substantially affected by this charge-control agent.

In a case where a negatively electrifiable photoconductive material is used for development with an oppositely electrifiable toner, or a positively electrifiable photoconductive material is used for reverse development, a positively electrifiable toner is used. In such a case, a positively electrifiable charge-control agent is used.

Further, in a case of a color toner, it is necessary to use a colorless charge-control agent or a charge-20 control agent with a pale color which does not affect the color of the toner. Such pale-colored or colorless charge-control agents may, for example, be quaternary ammonium salt compounds disclosed in e.g. Japanese Unexamined Patent Publication No. 119364/1982, No. 9154/1983 and No. 98742/1983.

However, these charge-control agents have drawbacks such that even when the toner has high electrifiability at the initial stage for the preparation of the developer, such electrifiability undergoes attenuation depending upon the storage conditions, and such attenuation tends to be remarkable especially when the temperature is high and the humidity is high. On the other hand, the p-halophenylcarboxylic acid disclosed in Japanese Unexamined Patent Publication No. 188752/1983 has a drawback that it is poor in the heat stability. Further, many of the above charge-control agents tend to provide oppositely electrifiable toners and have low electrifying effects. Otherwise, they have a drawback such that they are poor in the dispersibility or chemical stability. Thus, none of them has fully satisfactory properties as a charge-control agent.

It is an object of the present invention to provide a charge-control agent which has high stability as a colorless compound and good dispersibility to the binder resin and being free from a deterioration during the preparation of a toner and which is capable of presenting a toner which has a good tribocharge property and which is capable of constantly presenting an image of high image quality under various environmental conditions.

The present inventors have found a stable compound which has excellent dispersibility in a binder resin and which is capable of imparting an excellent tribocharge property to a toner, and have arrived at the present invention of an excellent toner by using this compound as a charge-control agent.

Namely, the present invention provides an electrostatic image developing toner containing a compound of the following formula (1):

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(wherein A is an electron attractive group, and each of R₁ and R₂ which are independent of each other, is a hydrogen atom, a halogen atom, an alkyl group, a cycloalkyl group, an alkoxy group, an aryl group, an

aralkyl group or a hydroxyl group, or R₁ and R₂ together form a ring),

(wherein A. R. and R. are as defined above), or

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20 (wherein A, R₁ and R₂ are as defined above), and Y is a hydrogen atom, an alkyl group or an aryl group. Now, the present invention will be described in detail with reference to the preferred embodiments.

Basically, the toner of the present invention comprises a binder resin, a coloring agent and the compound of the formula (1) of the present invention. As a method for producing the toner of the present invention, there may be mentioned a method wherein a misture of such starting materials are kneaded by a se heat-mixing apparatus while the binder resin is melted, and the mixture is then cooled, followed by rough pulverization, fine pulverization and classification, a method wherein a mixture of such starting materials is dissolved in a solvent and then sprayed to form fine particles, followed by drying and classification, or a method wherein the coloring agent and the compound of the formula (1) are dispersed in suspended monomer particles, followed by polymerization.

As the binder resin, a polystyrene, a styrene-methacrylate copolymer, a styrene-propylene copolymer, a styrene-butadiene copolymer, an acrylic resin, a styrene-maleic acid copolymer, an olefin resin, a polyester, an epoxy resin, a polyurethane resin, a polyvinyl butyral, etc., may be used alone or in combination as a mixture.

As the coloring agent, carbon black is commonly used for a black toner. For color toners, the following coloring agents are usually employed. Namely, as a yellow coloring agent, an azo-typo organic pigment such as CI pigment yellow 1, CI pigment yellow 5, CI pigment yellow 12 or CI pigment yellow 17, an inorganic pigment such as yellow oshre, or an oil-soluble dye such as CI solvent yellow 2, CI solvent yellow 19, may be mentioned. As a magenta coloring agent, an azo pigment such as CI pigment red 57 or CI pigment red 57:1, a xanthene pigment such as CI pigment violet 14 or CI pigment as CI pigment such as CI pigment as CI pigment violet 38, or an oil-soluble dye such as CI solvent red 49 or C

Such a coloring agent is used usually in an amount of from 1 to 15 parts by weight, preferably from 3 to 10 parts by weight, per 100 parts by weight of the binder resin.

The electron attractive group in the compound of the present invention useful as a charge-control agent, may, for example, be a fluorine atom, a chlorine atom, a thorine atom, a indiene atom, a indiene atom, a local soft atom, a tomical atom, a local soft atom, a tomical atom, a local soft atom, as for a creaming a control group, a cyano group, a fluoring group, a creative group, and a cyloxy group, an acyloxy group, and acyloxy group, and acyloxy group, and acyloxy group, as uptained as the group, and allysulfonyl group, as ubtained suffing group, as ubtained suffing group, as uptained as the group as a substituted suffing group, as uptained as the group as the group as group a

The following compounds may be mentioned as specific examples of the compound of the present invention useful as a charge-control agent.

(Compound No. 1)

(Compound No. 2)

(Compound No. 3)

$$I \longrightarrow OCH_2 COOP$$

(Compound No. 4)

$$NC \longrightarrow OCH_2COO$$

(Compound No. 5)

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(Compound No. 6)

$$H_3 C - C \longrightarrow OCH_2 COOH$$
 (Compound No. 7)

(Compound No. 8)

$$H_3 C - SO_2$$
 —OCH₂ COOH (Compound No. 9)

$$H_3$$
 C CH_3 $F \longrightarrow OCH_2$ COOH (Compound No. 19)

$$O_2 N - OCHCOOH$$
 (Compound No. 24)

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$$CH_2$$

C ℓ OCH_2 COOH (Compound No. 25)

O₂ N
$$\stackrel{\checkmark}{+}$$
 OCH₂ COOH (Compound No. 27)

Such a charge-control agent is used usually in an amount of from 0.1 to 10 parts by weight, preferably from 0.5 to 5 parts by weight, per 100 parts by weight of the binder resin.

The toner may further contain various additives such as hydrophobic silica, metal soap, a fluorine-type surfactant, dioctyl phthalate, wax, tin oxide and electrically conductive zinc oxide for the purposes of protecting the photoconductive material or carrier, improving the flowability of the toner, regulating the thermal properties, electrical properties and physical properties, regulating the electrical resistance, regulating the defining point and improving the fixing property.

When the toner of the present invention is used for a two-component developing agent, there may be employed, as a carrier, fine glass beads, iron powder, territe powder or a binder-type carrier of resin particles having magnetic particles dispersed therein, or a resin coated carrier having its surface coated with a polyester resin, a fluorine resin, an acrylic resin or a silicone resin. Further, the toner of the present invention exhibits excellent performance when used as a one-component toner.

Now, the present invention will be described in further detail with reference to Examples. However, it should be understood that the present invention is by no means restricted by such specific Examples. In the following Examples, "parts" means "parts by weight".

5 EXAMPLE 1

One part of p-fluorophenoxy acetic acid (Compound No. 1), 5 parts of carbon black and 94 parts of a styrene-ethylhexyl methacrylate copolymer were kneaded by a heat-mixing apparatus. After cooling, the mixture was roughly pulverized by a hammer mill, then finely pulverized by a jet mill and classified to to totain a black toner of from 10 to 12 µm. This toner was mixed with an iron powder carrier at a weight ratio of 4:100, and the mixture was shaked, whereby the toner was positively charged, and the tribocharge was measured by a blow off powder charge measuring apparatus and found to be +28 µc/g. This toner was used to copy an image by a modified commercially available copying machine, whereby copy images with an excellent image quality were obtained not only at the initial stage but also after coopying 10.000 sheetls.

EXAMPLE 2

One part of p-chlorophenoxy acetic acid (Compound No. 2), 5 parts of carbon black and 94 parts of a styrene-ethylhexyl methacrylate copolymer were kneaded by a heat-mixing apparatus. After cooling, the mixture was roughly pulvefized by a hammer mill, then finely pulverized by a jet mill and classified to obtain a black toner of from 10 to 12 µm. This toner was mixed with an iron powder carrier at a weight ratio of 4:100, and the mixture was shaked, whereby the toner was positively charged, and the tribcotrage measured by a blow off powder charge measuring apparatus was +23 µc/g. This toner was used to copy an image by a modified commercially available copying machine, whereby copy images with an excellent image quality were obtained not only at the initial stage but also after copying 10,000 sheets.

EXAMPLE 3

One part of p-cyanophenoxy acetic acid (Compound No. 5), 5 parts of Spilon Blue 2BNH as a copper phthalocyanine type oil-soluble dye (product of Hodogaya Chemical Co., Ltd.) and 94 parts of a styrene-butyl methacrylate copolymer were kneaded by a heat-mixing apparatus. After cooling, the mixture was roughly pulverized by a let mill and classified to obtain a blue toner of from 10 to 12 µm. This toner was mixed with an iron powder carrier at a weight ratio of 4:100, and the mixture was shaked, whereby the toner was positively charged, and the tribocharge measured by a blow off powder charge measuring apparatus was +20 µc/g. This toner was used to copy an image by a modified commercially available copying machine, whereby copy images with an excellent image quality were obtained not only at the initial stage but also after coopyring 10,000 sheets.

EXAMPLE 4

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One part of p-nitrophenoxy acetic acid (Compound No. 6), 5 parts of carbon black and 94 parts of a styrene-ethylhexyl methacrylate copolymer were kneaded by a heat-mixing apparatus. After cooling, the mixture was roughly pulverized by a hammer mill, then finely pulverized by a jet mill and classified to obtain a black toner of from 10 to 12 µm. This toner was mixed with a silicon resin coated carrier at a 46 weight ratio of 4:100, and the mixture was shaked, whereby the toner was positively charged, and the tribocharge measured by a blow off powder charge measuring apparatus was +18 µc/g. This toner was used to copy an image by a modified commercially available copying machine, whereby copy images with an excellent image quality were obtained not only at the initial stage but also after copying 10,000 sheetls.

50 EXAMPLE 5

One part of p-fluorophenoxy acetic acid (Compound No. 1), 40 parts of magnetic iron powder and 59 parts of a styrene-ethylinexyl methacytale copolymer were kneaded by a heat-mixing apparatus. After cooling, the mixture was roughly pulverized by a hammer mill, then finely pulverized by a jet mill and 50 classified to obtain a black toner of from 10 to 12 µm. This toner was mixed with a ferrite carrier, and the mixture was shaked, whereby the toner was positively charged. This toner was used to copy an image by a modified commercially available copying machine for one-component toner, whereby copy images with an excellent image quality were obtained.

EXAMPLES 7 TO 12

Experiments were conducted in the same manner as in Example 1 except that the compounds as identified in Table 1 were used instead of Compound No. 1 in Example 1, and the results are shown in 5 Table 1.

Table 1

E	Example No.	Compound No.	Tribo-charge of the toner (+ µc/g)		Image quality	
				Initial	After copying 10,000 sheets	
Г	6	Compound No. 7	35	Clear	Clear	
	7	Compound No. 10	20	Clear	Clear	
	8	Compound No. 14	15	Clear	Clear	
	9	Compound No. 19	12	Clear	Clear	
	10	Compound No. 24	11	Clear	Clear	
	11	Compound No. 26	17	Clear	Clear	

Claims

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1. An electrostatic image developing toner containing a compound of the following formula (1):

wherein X is a

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(wherein A is an electron attractive group, and each of R_1 and R_2 which are independent of each other, is a hydrogen atom, a halogen atom, an alkyl group, a cycloalkyl group, an alkoxy group, an anyl group, an aralkyl group or a hydroxyl group, or R_1 and R_2 together form a ringly.



(wherein A, R₁ and R₂ are as defined above), or

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$$\mathbf{A} - \underbrace{\bigcirc}^{\mathbf{R_1}} - \underbrace{\bigcirc}^{\mathbf{R_2}}$$

(wherein A, R_1 and R_2 are as defined above), and Y is a hydrogen atom, an alkyl group or an aryl group.

- The electrostatic image developing toner according to Claim 1, which comprises 100 parts by weight of a binder resin, from 1 to 15 parts by weight of a coloring agent and from 0.1 to 10 parts by weight of the compound of the formula (1).
- 3. The electrostatic image developing toner according to Claim 1, wherein the electron attractive group for A is a fluorine atom, a chlorine atom, a bromine atom, an iodine atom, a halogen-substituted alkyl group, a charbody group, a carbody group, a m N-substituted carbarnoyl group, an alkoyu carbonyl group, an acyl group, an arylcarbonyl group, a nitro group, a sullionic acid group, an alkylsulfonyl group, a substituted sulfamoyl group, or a substituted sulfamoyl group, a sulfamoyl group.
 - 4. The electrostatic image developing toner according to Claim 1, wherein the compound of the formula (1) is one of the following compounds:

$$C \ell - \bigcirc$$
 OCH₂ COOH (2)

$$Br - OCH_2 COOH$$
 (3)

$$NC \longrightarrow OCH_2COOH$$
 (5)

$$O_2 N - OCH_2 COOH$$
 (6)

$$H_3 C - C \longrightarrow OCH_2 COOH$$

$$F_3 C - \bigcirc - OCH_2 COOH$$
 (8)

$$H_3 C-SO_2 - \bigcirc -OCH_2 COOH$$
 (9)

$$F$$
 F
 OCH_2COOH (11)

$$CH_2$$
 HO_3 S $-OCH_2$ COOH (14)

$$F \xrightarrow{\hspace*{1cm}} OCH_2 COOH \tag{15}$$

F₃ C
$$\longrightarrow$$
 OCH₂ COOH (17)

$$F \longrightarrow OCH_2 COOH$$
 (18)

$$H_3$$
 C CH_3 F $-COOH$ $COOH$

$$O_2 N \longrightarrow OCH_2 COOH (20)$$

$$O_2 N - \bigcirc O_{COHCOOH}$$
 (24)

$$\begin{array}{c} \text{CH}_2 \\ \text{CPOCH}_2 \text{COOH} \end{array} \tag{25}$$

$$\begin{array}{c|c} \text{CH}_3 & \text{CH}_3 \\ \text{F} & \bigcirc & \text{OCHCOOH} \end{array} \quad (26)$$

$$O_2 N - H - OCH_2 COOH$$
 (27)



EUROPEAN SEARCH REPORT

EP 94 10 6686

	DOCUMENTS CONSIL				
Category	Citation of document with in of relevant pas	dication, where appropriate,	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.CL5)	
٨	PATENT ABSTRACTS OF vol. 11, no. 342 (P- November 1987 & JP-A-62 125 367 (I * abstract *	-635) (2789) 10	1	G03G9/097	
D,A	FR-A-2 524 991 (XERI * claim 1 *	(X)	1		
A	EP-A-0 490 370 (MIT	SUBISHI)	1		
				TECHNICAL FIELDS SEARCHED (Int.Cl.5)	
			i i	G03G	
	The present search report has b	een drawn up for all claims	-		
	Place of search	Date of completion of the search	1	Examiner	
	THE HAGUE	6 July 1994	Va	nhecke, H	
Y:po	CATEGORY OF CITED DOCUME articularly relevant if taken alone articularly relevant if combined with an ocument of the same category chnological background	E : earlier paten after the fili other D : document ci L : document ci	ted in the application ted for other reason	blished on, or on	
	on-written disclosure	& : member of t	he same patent fam	illy, corresponding	